

2118901 EPA NO. <u>U590002</u> FILE NO. <u>P2-13</u>

PROCESS FLOWCHART AND PORCESS DESCRIPTION

Attachment L

Information Request #6

6. PRODUCTION PLAN

6.1 Flowsheet and Process Description

The target end-product for the first year of operation will be cobalt sulfate solution containing 8 % cobalt.

The production of cobalt sulfate solution involves two key steps: dissolution of cobalt in sulfuric acid followed by purification by removing impurity metals. This last step is not necessary if virgin metal is used.

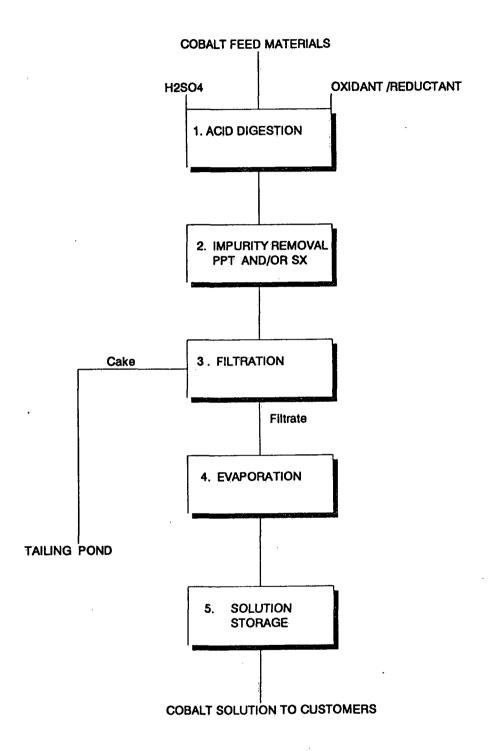
If a crystal product is desired at a later date, the additional steps of crystallization, drying and packaging will also be required.

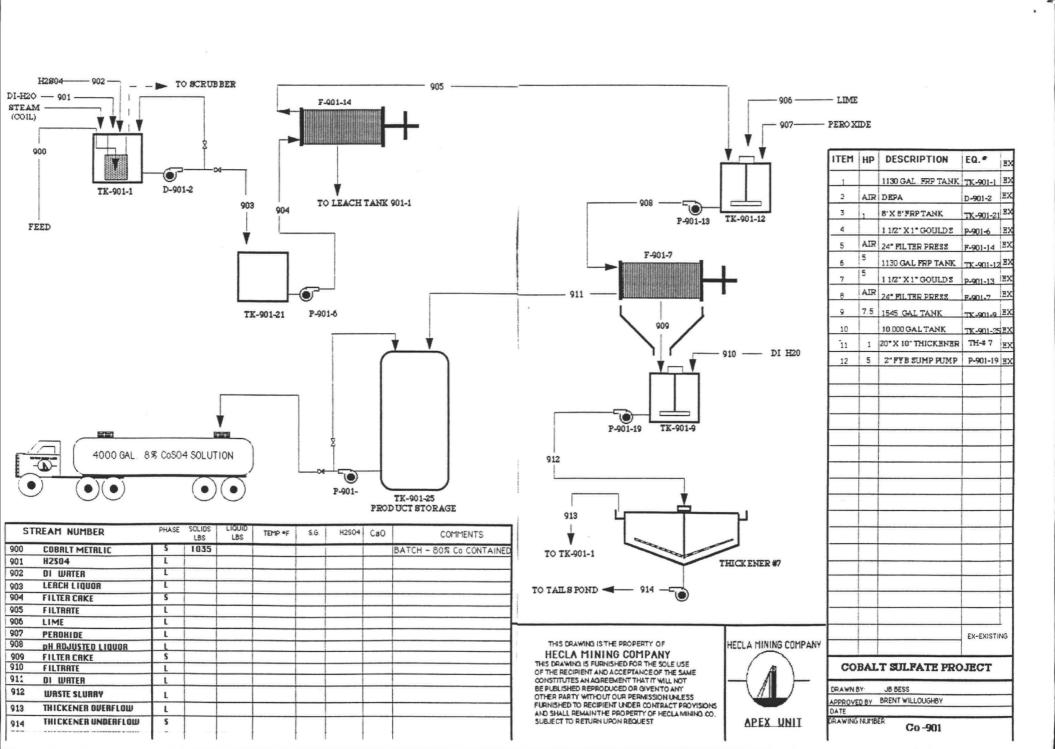
The process is shown by the block diagram flowsheet in Figures 6.1 and 6.2.

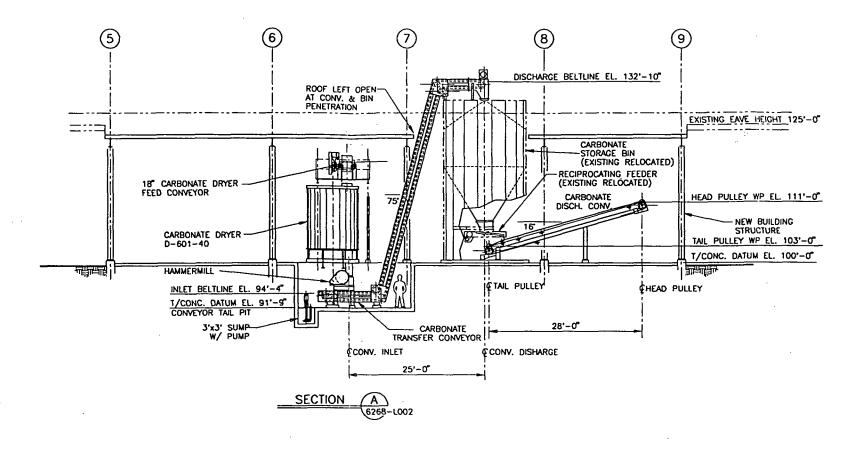
The dissolution rate of cobalt metal or cobalt oxide is influenced primarily by the oxidation state as well as the surface area of the feed material. An oxidant such as hydrogen peroxide or reductant, such as sugar is added to speed up the digestion. Generally the desirable cobalt concentration of the solution can be reached in less than 8 hours at an elevated temperature (80 to 100 oC). Dissolution difficulties have recently been encountered by some domestic cobalt sulfate producers with Russian cobalt ingots due to the low surface area of the material.

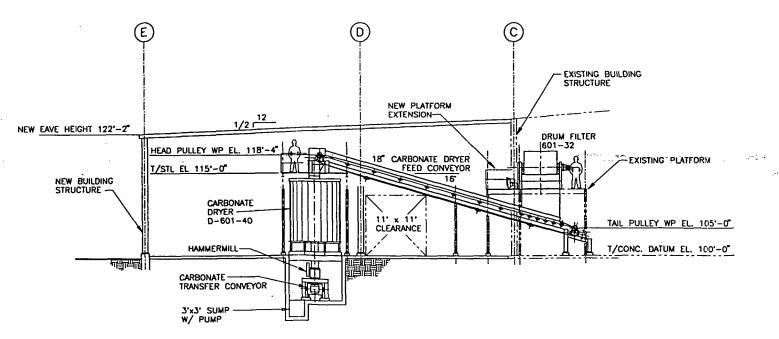
The impurity metals such as iron, aluminum and chromium are usually removed by precipitation at a Ph of 5.

A final adjustment of the solution strength is usually required before storage. This will be accomplished by either evaporation or dilution. There is little risk of crystallization or other product degradation during storage and shipment.









SECTION

<u>LEGEND</u>

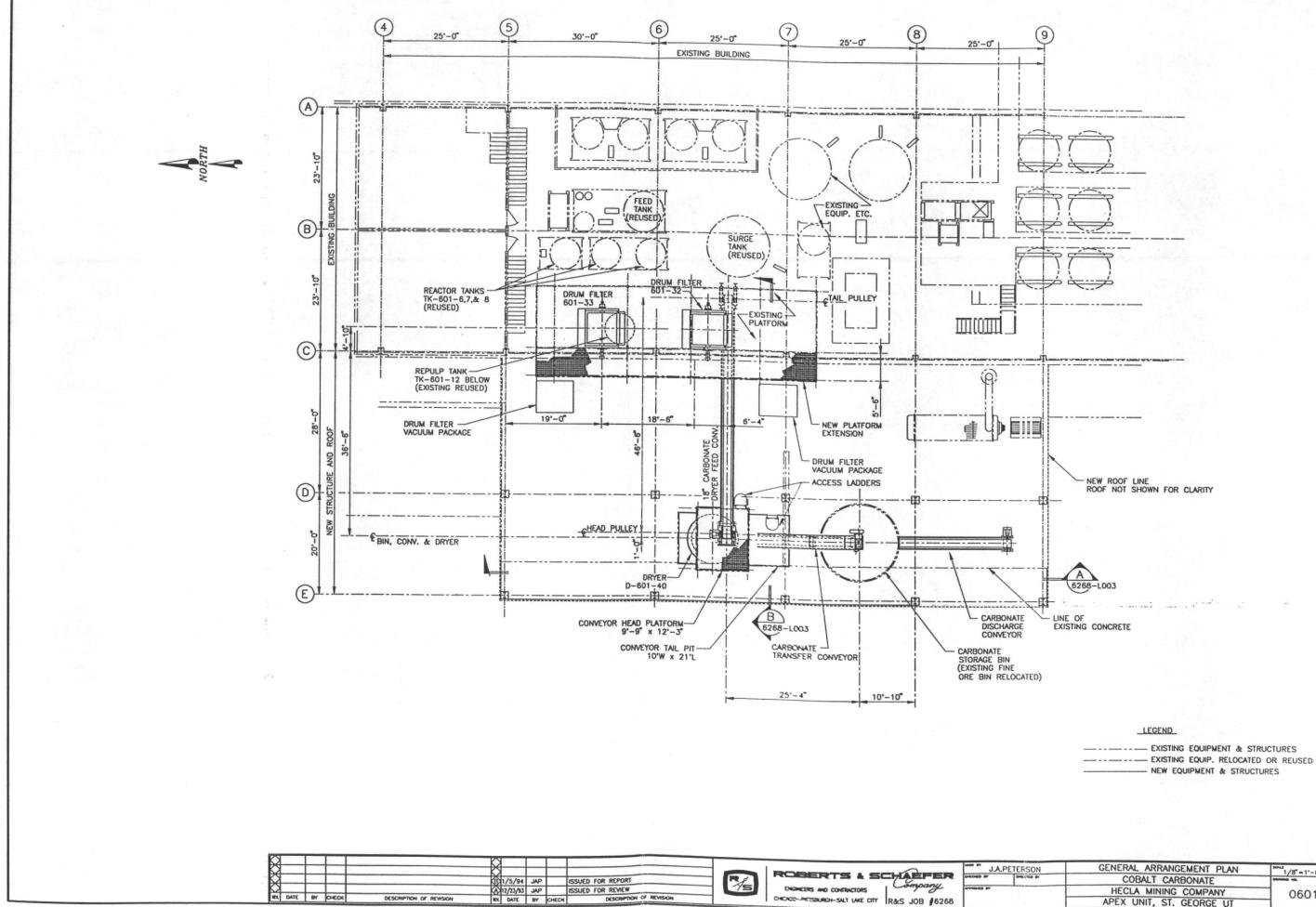
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